

Methyl 3-hydroxy-4-oxo-3,4-dihydro-2H-1,2-benzothiazine-3-carboxylate 1,1-dioxide monohydrate

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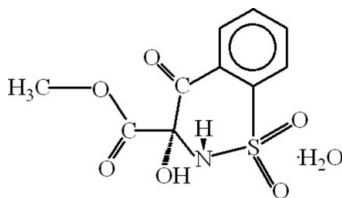
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.047; wR factor = 0.122; data-to-parameter ratio = 16.3.

In the molecule of the title compound, $\text{C}_{10}\text{H}_9\text{NO}_6\text{S}\cdot\text{H}_2\text{O}$, the benzothiazine ring adopts an envelope conformation. An intramolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond results in the formation of a nonplanar five-membered ring which has a twisted conformation. In the crystal structure, intermolecular $\text{N}-\text{H}\cdots\text{O}$, $\text{O}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules to form a three-dimensional network. There is a $\pi-\pi$ contact between the benzene rings [centroid-centroid distance = 3.972 (2) Å].

Related literature

For general background, see: Shafiq, Khan *et al.* (2008); Shafiq, Tahir *et al.* (2008); Tahir *et al.* (2008). For related literature, see: Antsyshkina *et al.* (2003); Allen (2002). For bond-length data, see: Allen *et al.* (1987). For ring puckering parameters, see: Cremer & Pople (1975).



Experimental

Crystal data

$\text{C}_{10}\text{H}_9\text{NO}_6\text{S}\cdot\text{H}_2\text{O}$
 $M_r = 289.26$
 Orthorhombic, $Pbca$
 $a = 7.7504$ (5) Å
 $b = 14.5638$ (9) Å
 $c = 21.0615$ (14) Å

$V = 2377.3$ (3) Å³
 $Z = 8$
 Mo $K\alpha$ radiation
 $\mu = 0.30$ mm⁻¹
 $T = 296$ (2) K
 $0.24 \times 0.18 \times 0.15$ mm

Data collection

Bruker Kappa APEXII CCD diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2005)
 $T_{\min} = 0.934$, $T_{\max} = 0.958$

14889 measured reflections
 2998 independent reflections
 1895 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.065$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.047$
 $wR(F^2) = 0.122$
 $S = 1.01$
 2998 reflections
 184 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.40$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.32$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{O6}$	0.78 (3)	2.43 (3)	2.744 (3)	106 (2)
$\text{N1}-\text{H1}\cdots\text{O7}^i$	0.78 (3)	2.29 (3)	3.032 (3)	162 (3)
$\text{O4}-\text{H4O}\cdots\text{O7}^{ii}$	0.84 (3)	1.94 (3)	2.773 (3)	175 (2)
$\text{O7}-\text{H71}\cdots\text{O3}^{iii}$	0.83 (3)	2.45 (3)	3.107 (3)	137 (3)
$\text{O7}-\text{H71}\cdots\text{O5}^{iv}$	0.83 (3)	2.45 (3)	3.028 (3)	128 (3)
$\text{O7}-\text{H72}\cdots\text{O2}$	0.83 (4)	2.21 (4)	3.027 (3)	167 (3)
$\text{C5}-\text{H5}\cdots\text{O4}^{iv}$	0.9300	2.4800	3.374 (3)	162.00
$\text{C10}-\text{H10A}\cdots\text{O4}^i$	0.9600	2.3100	2.994 (3)	128.00

Symmetry codes: (i) $x-1, y, z$; (ii) $x-\frac{1}{2}, -y+\frac{1}{2}, -z$; (iii) $-x+1, -y, -z$; (iv) $-x+\frac{1}{2}, y-\frac{1}{2}, z$.

Data collection: APEX2 (Bruker, 2007); cell refinement: APEX2; data reduction: SAINT (Bruker, 2007); program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and PLATON (Spek, 2003); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HK2537).

References

- Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.
 Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
 Antsyshkina, A. S., Sadikov, G. G., Korshunov, O. Yu., Anpilova, E. L., Bicherov, A. S., Sergienko, V. S., Uflyand, I. E. & Garnovskii, A. D. (2003). *Russ. J. Coord. Chem.* **29**, 724–731.
 Bruker (2005). SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (2007). APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
 Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
 Shafiq, M., Khan, I. U., Tahir, M. N. & Siddiqui, W. A. (2008). *Acta Cryst.* **E64**, o558.
 Shafiq, M., Tahir, M. N., Khan, I. U., Ahmad, S. & Siddiqui, W. A. (2008). *Acta Cryst.* **E64**, o1270.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
 Tahir, M. N., Shafiq, M., Khan, I. U., Siddiqui, W. A. & Arshad, M. N. (2008). *Acta Cryst.* **E64**, o557.

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Acta Cryst. (2008). E64, o2045 [doi:10.1107/S1600536808030948]

Methyl 3-hydroxy-4-oxo-3,4-dihydro-2H-1,2-benzothiazine-3-carboxylate 1,1-dioxide monohydrate

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Comment

The title compound has been prepared in continuation of research on benzo-thiazine derivatives (Shafiq, Khan *et al.*, 2008; Shafiq, Tahir *et al.*, 2008; Tahir *et al.*, 2008) by our research group. The CCDC search (Allen, 2002) shows that a single crystal structure has been reported, in which the same benzothiazine ring exists (Antsyshkina *et al.*, 2003). The title compound differs from the reported structure, due to the hydroxy and methylformate groups. Due to the hydroxy group, the S-configuration in the title compound has been confirmed.

In the molecule of the title compound (Fig. 1), the bond lengths (Allen *et al.*, 1987) and angles are within normal ranges. Ring A (C1-C6) is, of course, planar. Ring B (S1/N1/C1/C6-C8) is not planar, having total puckering amplitude, Q_T , of 0.733 (3) Å and envelope conformation [$\varphi = 21.21$ (3)° and $\theta = 76.70$ (3)°] (Cremer & Pople, 1975) with N1 atom displaced by 0.575 (3) Å from the plane of the other ring atoms. The intramolecular N-H...O hydrogen bond (Table 1) results in the formation of a nonplanar five-membered ring C (N1/O6/C8/C9/H1), having twisted conformation.

In the crystal structure, intermolecular N-H...O, O-H...O and C-H...O hydrogen bonds (Table 1) link the molecules to form a three dimensional network (Fig. 2), in which they may be effective in the stabilization of the structure. The π — π contact between the benzene rings, $Cg2 \cdots Cg2^i$ [symmetry code: (i) $-1/2 + x, y, 1/2 - z$, where $Cg2$ is the centroid of the ring A (C1-C6)] may further stabilize the structure, with centroid-centroid distance of 3.972 (2) Å.

Experimental

For the preparation of the title compound, methyl 4-hydroxy-2H-1,2-benzothiazine-3-carboxylate 1,1-dioxide (0.5 g, 1.95 mmol), *N*-bromo-succinamide (0.38 g, 2.145 mmol) and dibenzoyl peroxide (0.035 g, 0.15 mmol) were added in CCl_4 (10 ml). The mixture was refluxed for 2 h. After the completion of reaction, CCl_4 was distilled off under vacuum. The obtained residue was washed with hot water to remove other impurities. The solid product was recrystallized in water and methanol to obtain the suitable crystals for x-ray analysis.

Refinement

H atoms were located in difference syntheses and refined as [O-H = 0.84 (3) Å (for OH); O-H = 0.83 (3) and 0.83 (4) Å (for H₂O); N-H = 0.78 (3) Å (for NH)]. The remaining H atoms were positioned geometrically, with C-H = 0.93 and 0.96 Å for aromatic and methyl H, respectively, and constrained to ride on their parent atoms with $U_{iso}(H) = xU_{eq}(C,N,O)$, where $x = 1.5$ for methyl H and $x = 1.2$ for all other H atoms.

Figures

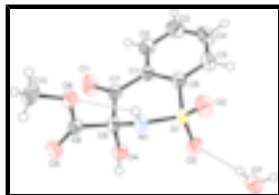


Fig. 1. The molecular structure of the title molecule, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. Hydrogen bonds are shown as dotted lines.

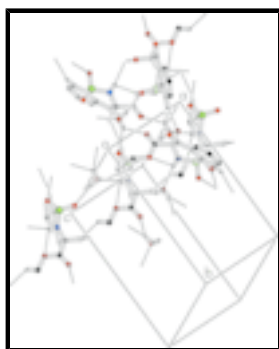


Fig. 2. A packing diagram of the title compound. Hydrogen bonds are shown as dashed lines.

Methyl 3-hydroxy-4-oxo-3,4-dihydro-2H-1,2-benzothiazine-3- carboxylate 1,1-dioxide monohydrate

Crystal data

$C_{10}H_9NO_6S \cdot H_2O$

$M_r = 289.26$

Orthorhombic, *Pbca*

Hall symbol: -P 2ac 2ab

$a = 7.7504$ (5) Å

$b = 14.5638$ (9) Å

$c = 21.0615$ (14) Å

$V = 2377.3$ (3) Å³

$Z = 8$

$F_{000} = 1200$

$D_x = 1.616$ Mg m⁻³

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 2998 reflections

$\theta = 3.0$ – 28.5°

$\mu = 0.30$ mm⁻¹

$T = 296$ (2) K

Prismatic, colourless

$0.24 \times 0.18 \times 0.15$ mm

Data collection

Bruker Kappa APEXII CCD diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

Detector resolution: 7.40 pixels mm⁻¹

$T = 296$ (2) K

ω scans

Absorption correction: multi-scan (SADABS; Bruker, 2005)

$T_{\min} = 0.934$, $T_{\max} = 0.958$

14889 measured reflections

2998 independent reflections

1895 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.065$

$\theta_{\text{max}} = 28.5^\circ$

$\theta_{\text{min}} = 3.0^\circ$

$h = -10 \rightarrow 10$

$k = -19 \rightarrow 10$

$l = -28 \rightarrow 27$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.047$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.122$	$w = 1/[\sigma^2(F_o^2) + (0.0491P)^2 + 1.1022P]$
$S = 1.01$	where $P = (F_o^2 + 2F_c^2)/3$
2998 reflections	$(\Delta/\sigma)_{\max} < 0.001$
184 parameters	$\Delta\rho_{\max} = 0.40 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	$\Delta\rho_{\min} = -0.32 \text{ e } \text{\AA}^{-3}$
	Extinction correction: none

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.18084 (8)	0.12546 (4)	0.08206 (3)	0.0314 (2)
O1	0.0037 (3)	0.32676 (13)	0.21865 (9)	0.0568 (8)
O2	0.3475 (2)	0.15746 (13)	0.06338 (9)	0.0450 (7)
O3	0.1137 (3)	0.04539 (12)	0.05197 (9)	0.0450 (6)
O4	0.2024 (2)	0.33701 (12)	0.09947 (10)	0.0360 (6)
H4O	0.204 (4)	0.3501 (19)	0.0608 (14)	0.0432*
O5	-0.0985 (3)	0.42862 (12)	0.08273 (10)	0.0460 (7)
O6	-0.2498 (2)	0.30222 (13)	0.10613 (9)	0.0413 (6)
O7	0.7145 (3)	0.10998 (15)	0.02689 (11)	0.0481 (7)
H71	0.731 (4)	0.054 (2)	0.0226 (15)	0.0577*
H72	0.615 (5)	0.116 (2)	0.0412 (16)	0.0577*
N1	0.0398 (3)	0.20645 (14)	0.07099 (10)	0.0308 (6)
H1	-0.052 (4)	0.1857 (18)	0.0675 (13)	0.0370*
C1	0.1243 (3)	0.17959 (16)	0.20533 (12)	0.0298 (7)
C2	0.1294 (4)	0.16328 (19)	0.27020 (13)	0.0414 (9)
H2	0.09091	0.20830	0.29812	0.0497*
C3	0.1908 (4)	0.0814 (2)	0.29384 (14)	0.0518 (10)
H3	0.19348	0.07164	0.33746	0.0623*

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C4	0.2479 (5)	0.0141 (2)	0.25324 (15)	0.0525 (10)
H4	0.28955	-0.04093	0.26959	0.0629*
C5	0.2440 (4)	0.02761 (18)	0.18834 (14)	0.0444 (9)
H5	0.28239	-0.01804	0.16088	0.0533*
C6	0.1824 (3)	0.10985 (16)	0.16477 (12)	0.0306 (7)
C7	0.0552 (3)	0.26875 (16)	0.18234 (12)	0.0319 (8)
C8	0.0514 (3)	0.28972 (15)	0.11076 (11)	0.0284 (7)
C9	-0.1075 (3)	0.35000 (17)	0.09736 (12)	0.0312 (8)
C10	-0.4122 (3)	0.3507 (2)	0.10026 (15)	0.0500 (10)
H10A	-0.50577	0.30900	0.10785	0.0750*
H10B	-0.42207	0.37576	0.05825	0.0750*
H10C	-0.41623	0.39963	0.13080	0.0750*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0315 (3)	0.0292 (3)	0.0334 (3)	0.0025 (3)	0.0002 (3)	-0.0061 (3)
O1	0.0826 (16)	0.0444 (11)	0.0433 (12)	0.0237 (12)	-0.0002 (11)	-0.0134 (9)
O2	0.0328 (11)	0.0479 (11)	0.0544 (12)	0.0054 (9)	0.0128 (9)	-0.0020 (9)
O3	0.0591 (13)	0.0327 (9)	0.0433 (11)	0.0012 (9)	-0.0078 (10)	-0.0121 (8)
O4	0.0240 (9)	0.0376 (10)	0.0463 (11)	-0.0060 (8)	-0.0014 (8)	-0.0029 (9)
O5	0.0406 (12)	0.0312 (10)	0.0662 (14)	0.0041 (8)	-0.0023 (10)	0.0090 (9)
O6	0.0220 (9)	0.0402 (10)	0.0618 (13)	-0.0003 (8)	0.0007 (9)	0.0062 (9)
O7	0.0453 (13)	0.0400 (10)	0.0590 (14)	-0.0025 (10)	-0.0070 (10)	0.0005 (10)
N1	0.0258 (11)	0.0296 (10)	0.0370 (12)	-0.0008 (9)	-0.0049 (10)	-0.0052 (9)
C1	0.0261 (12)	0.0291 (12)	0.0342 (13)	0.0004 (10)	-0.0018 (10)	-0.0021 (10)
C2	0.0470 (16)	0.0411 (15)	0.0362 (14)	0.0006 (13)	0.0007 (13)	-0.0049 (12)
C3	0.063 (2)	0.0559 (18)	0.0364 (15)	0.0002 (17)	-0.0064 (15)	0.0089 (14)
C4	0.067 (2)	0.0393 (15)	0.0513 (18)	0.0054 (15)	-0.0086 (16)	0.0110 (14)
C5	0.0503 (17)	0.0342 (14)	0.0487 (17)	0.0069 (13)	-0.0052 (14)	-0.0021 (13)
C6	0.0269 (12)	0.0317 (12)	0.0332 (13)	0.0008 (11)	-0.0037 (11)	-0.0016 (10)
C7	0.0284 (13)	0.0296 (12)	0.0377 (14)	0.0005 (10)	-0.0013 (11)	-0.0055 (11)
C8	0.0240 (12)	0.0257 (11)	0.0356 (14)	-0.0001 (10)	-0.0027 (10)	-0.0041 (10)
C9	0.0272 (13)	0.0331 (13)	0.0332 (14)	-0.0001 (11)	-0.0003 (10)	-0.0017 (10)
C10	0.0234 (14)	0.0674 (19)	0.0591 (19)	0.0118 (14)	-0.0015 (13)	0.0030 (16)

Geometric parameters (\AA , $^\circ$)

S1—O2	1.4284 (17)	C1—C7	1.486 (3)
S1—O3	1.426 (2)	C1—C6	1.402 (3)
S1—N1	1.625 (2)	C2—C3	1.377 (4)
S1—C6	1.757 (3)	C3—C4	1.374 (4)
O1—C7	1.207 (3)	C4—C5	1.381 (4)
O4—C8	1.379 (3)	C5—C6	1.382 (4)
O5—C9	1.188 (3)	C7—C8	1.539 (3)
O6—C9	1.317 (3)	C8—C9	1.539 (3)
O6—C10	1.449 (3)	C2—H2	0.9300
O4—H4O	0.84 (3)	C3—H3	0.9300
O7—H71	0.83 (3)	C4—H4	0.9300

O7—H72	0.83 (4)	C5—H5	0.9300
N1—C8	1.477 (3)	C10—H10C	0.9600
N1—H1	0.78 (3)	C10—H10A	0.9600
C1—C2	1.387 (4)	C10—H10B	0.9600
S1…H10B ⁱ	3.0600	N1…O6	2.744 (3)
O1…O4	2.949 (3)	N1…O7 ^{xii}	3.032 (3)
O1…O6	3.099 (3)	C3…C5 ^x	3.570 (4)
O1…C4 ⁱⁱ	3.406 (4)	C4…O1 ^{ix}	3.418 (4)
O1…C4 ⁱⁱⁱ	3.418 (4)	C4…O1 ^{xiii}	3.405 (4)
O2…O4	2.946 (3)	C5…C3 ^{xi}	3.570 (4)
O2…O7	3.027 (3)	C5…O4 ^{ix}	3.374 (3)
O2…C9 ⁱ	3.405 (3)	C9…O2 ^{vii}	3.405 (3)
O3…C10 ^{iv}	3.393 (3)	C10…O4 ^{xii}	2.994 (3)
O3…O7 ^v	3.107 (3)	C10…O3 ^{xiv}	3.393 (3)
O3…O3 ^{vi}	3.106 (3)	C10…H4O ^{xii}	3.09 (3)
O4…O5	2.710 (3)	C10…H2 ^x	2.9800
O4…C5 ⁱⁱⁱ	3.374 (3)	H1…O6	2.43 (3)
O4…O7 ^{vii}	2.773 (3)	H1…O7 ^{xii}	2.29 (3)
O4…O2	2.946 (3)	H2…H10A ^{xi}	2.5800
O4…O1	2.949 (3)	H2…O1	2.5000
O4…C10 ^{viii}	2.994 (3)	H2…O6 ^{xi}	2.7300
O5…O7 ⁱⁱⁱ	3.028 (3)	H2…C10 ^{xi}	2.9800
O5…O4	2.710 (3)	H3…O7 ^x	2.9200
O6…N1	2.744 (3)	H3…O5 ^{xiii}	2.7800
O6…O1	3.099 (3)	H4…H10C ^{xiii}	2.4700
O7…N1 ^{viii}	3.032 (3)	H4…O1 ^{ix}	2.7300
O7…O5 ^{ix}	3.028 (3)	H4O…C10 ^{viii}	3.09 (3)
O7…O4 ⁱ	2.773 (3)	H4O…H10A ^{viii}	2.5300
O7…O3 ^v	3.107 (3)	H4O…O5	2.65 (3)
O7…O2	3.027 (3)	H4O…O7 ^{vii}	1.94 (3)
O1…H4 ⁱⁱⁱ	2.7300	H4O…H71 ^{vii}	2.25 (4)
O1…H2	2.5000	H4O…H72 ^{vii}	2.31 (4)
O2…H72	2.21 (4)	H5…O3	2.8000
O2…H10A ^{viii}	2.6500	H5…O4 ^{ix}	2.4800
O3…H10B ⁱ	2.6000	H10A…O2 ^{xii}	2.6500
O3…H5	2.8000	H10A…O4 ^{xii}	2.3100
O3…H71 ^v	2.45 (3)	H10A…H2 ^x	2.5800
O3…H10B ^{iv}	2.8900	H10A…H4O ^{xii}	2.5300
O4…H5 ⁱⁱⁱ	2.4800	H10B…S1 ^{vii}	3.0600
O4…H10A ^{viii}	2.3100	H10B…O3 ^{vii}	2.6000
O5…H4O	2.65 (3)	H10B…O5	2.6700
O5…H10B	2.6700	H10B…O3 ^{xiv}	2.8900

supplementary materials

O5...H3 ⁱⁱ	2.7800	H10C...O5	2.7000
O5...H72 ⁱⁱⁱ	2.87 (3)	H10C...H4 ⁱⁱ	2.4700
O5...H10C	2.7000	H71...H4O ⁱ	2.25 (4)
O5...H71 ⁱⁱⁱ	2.45 (3)	H71...O3 ^v	2.45 (3)
O6...H2 ^x	2.7300	H71...O5 ^{ix}	2.45 (3)
O6...H1	2.43 (3)	H72...O2	2.21 (4)
O7...H1 ^{viii}	2.29 (3)	H72...H4O ⁱ	2.31 (5)
O7...H4O ⁱ	1.94 (3)	H72...O5 ^{ix}	2.87 (3)
O7...H3 ^{xi}	2.9200		
O2—S1—O3	118.33 (12)	O1—C7—C8	118.4 (2)
O2—S1—N1	109.39 (11)	O4—C8—N1	111.36 (19)
O2—S1—C6	108.01 (11)	O4—C8—C7	104.59 (19)
O3—S1—N1	106.53 (12)	O4—C8—C9	111.27 (18)
O3—S1—C6	109.72 (11)	N1—C8—C7	113.20 (18)
N1—S1—C6	103.94 (11)	N1—C8—C9	108.41 (19)
C9—O6—C10	117.3 (2)	C7—C8—C9	107.96 (19)
C8—O4—H4O	107 (2)	O5—C9—C8	123.4 (2)
H71—O7—H72	107 (3)	O6—C9—C8	110.1 (2)
S1—N1—C8	118.28 (17)	O5—C9—O6	126.5 (2)
C8—N1—H1	115 (2)	C3—C2—H2	120.00
S1—N1—H1	110 (2)	C1—C2—H2	120.00
C2—C1—C6	117.8 (2)	C2—C3—H3	120.00
C2—C1—C7	118.7 (2)	C4—C3—H3	120.00
C6—C1—C7	123.4 (2)	C5—C4—H4	120.00
C1—C2—C3	120.9 (3)	C3—C4—H4	120.00
C2—C3—C4	120.3 (3)	C4—C5—H5	120.00
C3—C4—C5	120.5 (3)	C6—C5—H5	120.00
C4—C5—C6	119.1 (3)	O6—C10—H10B	109.00
S1—C6—C5	118.1 (2)	O6—C10—H10C	109.00
S1—C6—C1	120.56 (18)	O6—C10—H10A	110.00
C1—C6—C5	121.4 (2)	H10A—C10—H10C	109.00
O1—C7—C1	121.6 (2)	H10B—C10—H10C	109.00
C1—C7—C8	120.0 (2)	H10A—C10—H10B	110.00
O2—S1—N1—C8	-67.0 (2)	C2—C1—C7—C8	178.7 (2)
O3—S1—N1—C8	163.98 (18)	C6—C1—C7—O1	178.8 (2)
C6—S1—N1—C8	48.1 (2)	C6—C1—C7—C8	-2.4 (3)
O2—S1—C6—C1	94.5 (2)	C1—C2—C3—C4	-0.2 (5)
O2—S1—C6—C5	-84.8 (2)	C2—C3—C4—C5	-0.2 (5)
O3—S1—C6—C1	-135.2 (2)	C3—C4—C5—C6	0.2 (5)
O3—S1—C6—C5	45.5 (2)	C4—C5—C6—S1	179.4 (2)
N1—S1—C6—C1	-21.6 (2)	C4—C5—C6—C1	0.1 (4)
N1—S1—C6—C5	159.1 (2)	O1—C7—C8—O4	84.6 (3)
C10—O6—C9—O5	-3.0 (4)	O1—C7—C8—N1	-154.0 (2)
C10—O6—C9—C8	175.4 (2)	O1—C7—C8—C9	-33.9 (3)
S1—N1—C8—O4	64.3 (2)	C1—C7—C8—O4	-94.2 (2)
S1—N1—C8—C7	-53.3 (3)	C1—C7—C8—N1	27.2 (3)
S1—N1—C8—C9	-173.01 (16)	C1—C7—C8—C9	147.2 (2)

C6—C1—C2—C3	0.5 (4)	O4—C8—C9—O5	-4.0 (3)
C7—C1—C2—C3	179.4 (3)	O4—C8—C9—O6	177.6 (2)
C2—C1—C6—S1	-179.7 (2)	N1—C8—C9—O5	-126.8 (3)
C2—C1—C6—C5	-0.4 (4)	N1—C8—C9—O6	54.8 (3)
C7—C1—C6—S1	1.4 (3)	C7—C8—C9—O5	110.2 (3)
C7—C1—C6—C5	-179.3 (2)	C7—C8—C9—O6	-68.2 (2)
C2—C1—C7—O1	-0.1 (4)		

Symmetry codes: (i) $x+1/2, -y+1/2, -z$; (ii) $-x, y+1/2, -z+1/2$; (iii) $-x+1/2, y+1/2, z$; (iv) $-x-1/2, y-1/2, z$; (v) $-x+1, -y, -z$; (vi) $-x, -y, -z$; (vii) $x-1/2, -y+1/2, -z$; (viii) $x+1, y, z$; (ix) $-x+1/2, y-1/2, z$; (x) $x-1/2, y, -z+1/2$; (xi) $x+1/2, y, -z+1/2$; (xii) $x-1, y, z$; (xiii) $-x, y-1/2, -z+1/2$; (xiv) $-x-1/2, y+1/2, z$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 \cdots O6	0.78 (3)	2.43 (3)	2.744 (3)	106 (2)
N1—H1 \cdots O7 ^{xii}	0.78 (3)	2.29 (3)	3.032 (3)	162 (3)
O4—H4O \cdots O7 ^{vii}	0.84 (3)	1.94 (3)	2.773 (3)	175 (2)
O7—H71 \cdots O3 ^v	0.83 (3)	2.45 (3)	3.107 (3)	137 (3)
O7—H71 \cdots O5 ^{ix}	0.83 (3)	2.45 (3)	3.028 (3)	128 (3)
O7—H72 \cdots O2	0.83 (4)	2.21 (4)	3.027 (3)	167 (3)
C5—H5 \cdots O4 ^{ix}	0.9300	2.4800	3.374 (3)	162.00
C10—H10A \cdots O4 ^{xii}	0.9600	2.3100	2.994 (3)	128.00

Symmetry codes: (xii) $x-1, y, z$; (vii) $x-1/2, -y+1/2, -z$; (v) $-x+1, -y, -z$; (ix) $-x+1/2, y-1/2, z$.

Fig. 1

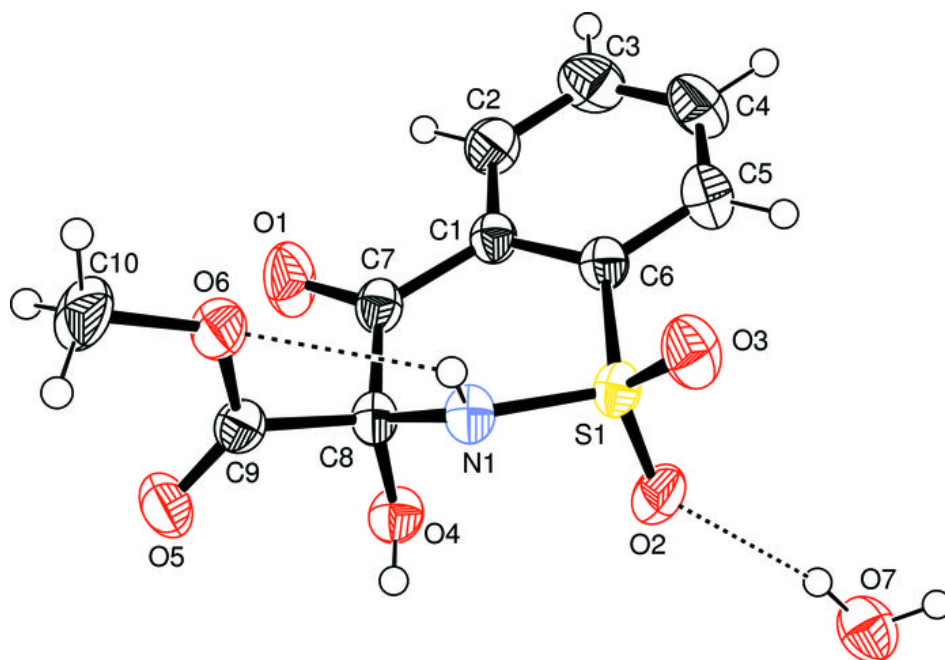


Fig. 2

